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Chapter 17 Thermochemistry 429 Section Review Objectives • Explain the relationship between energy, heat, and work • Distinguish between exothermic and endothermic processes • Distinguish between heat capacity and specific heat Vocabulary Key Equations and Relationships • 1 Calorie 1 kilocalorie 1000 calories • 1 J 0.2390 cal and 4.184 J 1 cal • C m

05 CTR ch17 7/12/04 8:15 AM Page 429 THE FLOW OF ENERGY ...

Chapter 17 Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505-510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

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Concept Review with Key Terms. 6.1 Energy—Common forms of energy include kinetic energy and potential energy.The SI unit of energy is the joule (J).When a force causes an object to move, work is done: Work = force × distance. 6.2 Thermochemistry: Some Basic Terms— Thermochemistry is the study of energy changes in physical processes or chemical reactions.

Thermochemistry - Pearson Education

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17. Thermochemistry, Heat, and Chemical Change 18. Reaction Rates and Equilibrium 19. Acids and Bases and Salts 20. Oxidation-Reduction Reactions 21. Electrochemistry 22. Hydrocarbon Compounds 23. Functional Groups and Organic Reactions 24. The Chemistry of Life 25. Nuclear Chemistry

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Chapter 12 - Stoichiometry - 12 Assessment - Page 413: 64

Chapter 17 Thermochemistry 431 Section Review Objectives • Construct equations that show the enthalpy changes for chemical and physical processes • Calculate enthalpy changes in chemical and physical processes Vocabulary Key Equation • q sys^H # q surr^H # C ΔT, where "T i Part ACompletion Use this completion exercise to check your understanding of the concepts and terms